Work, energy, and machines

Task 1

Work, energy, and machines

Work is done when a force moves an object and energy is transferred from one store to another.

1 The table has some examples of work being done.

In each case, state the force and the distance.

The first one has been done for you.

Action	Force	Distance
box pushed along the floor	friction	distance the box is moved
weightlifter lifting a weight		
catapult pulled back		
car crashes, leaving a dent		

2	State two examples of simple machines in everyday use.
3	Machines make work easier. Draw diagrams to illustrate and explain the following statements.
	a A screwdriver makes it easier to take the lid off a paint can.

b A spanner with a longer handle is ea	sier to use than one with a s	horter handle.
c A pulley system is used to lift a lifeb	oat.	
Task 2		
Thermal energy		
An object's thermal energy depends on it of.	s mass, temperature, and wh	at it is made
Read these statements and decide wheth	er they are true or false.	
Thermal energy is the same as temperate	ıre.	T / F
Some materials heat up more quickly tha	n others.	T / F
A full kettle of water takes longer that a h	nalf-filled kettle to boil.	T/F

T/F

Ice cubes cool drinks by transferring cold energy to the warmer drink.

Task 3

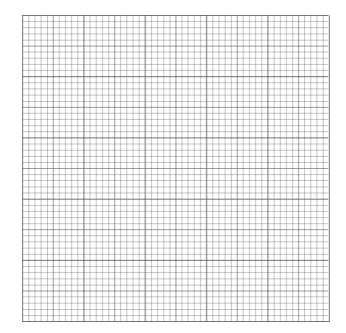
Investigating cooling

Etta investigates whether the starting temperature of a liquid affects the rate of cooling. The temperature change will be recorded for 200 cm³ of water over a 5-minute period for different starting temperatures. These are her results.

Starting temp. (°C)	Final temp. (°C)	Temp. fall (°C)
80	70	
75	66	
70	62	
65	59	
60	55	

1	Identify the:	
	independent variable	
	•	
	dependent variable	
	control variables.	
	Control variables.	

- **2** Calculate the temperature fall values, and complete the table above.
- **3** Plot a graph of *temperature fall* against *starting temperature*.



_	ask 4 hermal energy transfer by particles
	What temperature will all of the water be after an hour? Explain your answer.
	·
5	Room temperature was 20 °C.
	so, describe the relationship.
4	Do the results show a link between starting temperature and temperature fail? If

1	Draw the	arrangement	of particles	in a	a solid,	liquid,	and gas.
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Solid	Liquid	Gas

2 For each row, tick all boxes where you think the description is of particles in a solid, liquid, or gas.

Description of particles	Solid	Liquid	Gas
close together			
arranged in a pattern			
can move around each other			

3 Draw a diagram to show heat passing through a metal rod. Describe, using these key words, how the thermal energy is transferred.

vibrate	faster	particles	electrons	conduction	collide	thermal energy	
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4 The heating element in a kettle is always at the bottom.

Add to the diagram of a kettle to show how all the water in the kettle heats up, rather than just the bottom layer.

Use these key words.

	sink	dense	particles	rise	faster	convection current	
ket wa	tie ————————————————————————————————————						
hea ele	ating — — ment						

Task 5

shadow

Thermal energy transfer by radiation

straight line

If you are standing in front of a bonfire on a cold night and someone walks between you and the fire, you can no longer see the fire and instantly feel cold. Draw a diagram, and explain these observations using the following key words.

radiation

thermal

wave

							
Tas	k 6						
[nsu	lation						
Each	of these	questions	has a set o	of key words	to use in y	our answer.	
L Ex	plain wh	y a bird fluf	fs out its f	eathers in w	vinter to kee	ep warm.	
	trap	insulator	particles	far apar	t conduc	ctor convect	tion transfer
	plain wh the race		runners a	are given a s	hiny metall	c sheet to we	ear at the end
						c sheet to we	٦
						1	٦
						1	٦
of :	the race	rac	diates	reflects	cool	hot	٦

3	Explain why flasks used to keep liquids hot or cold have a vacuum layer.							
		vacuum	conduction	convection	radiation	particles	wave	
	_							
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