Matter

4. Elements

CONCEPT 3

TEST YOURSELF

CERAMICS, POLYMERS AND COMPOSITES



- Q1 Polymers can be broken down to the molecules they are made from. Is this a physical or a chemical change?
- Q2 What are the similarities between synthetic polymers and natural polymers?
- Q3 List 3 items made from ceramics in your home.
- Q4 Ceramic materials have been uncovered from earliest human history. What does this tell you about the nature of ceramics?



- Q5 Why is it useful to store small molecules (such as glucose) in the form of polymers (such as starch)?
- Q6 Plants store sugar in the form of starch; animals store it in the form of glycogen. What prediction(s) can you make about glycogen?
- Q7 Describe some examples of uses of synthetic polymers in place of metals.
- Q8 Draw a table to compare the properties of ceramics with metals.
- Glass and carbon fibres are strong and lightweight. What makes glass and carbon fibres a Q9 popular choice as reinforcers?



- Q10 Do you think that polymers are chemically the same as the monomers that make them? Explain your answer.
- Suggest why ceramics Q11 may be chosen over metals to make turbo-jet engine blades.

Type of material	Material	Density (g/cm³)	Strength (MPa)*	Strength/ weight ratio
composite	fibreglass	1.9	3400	1307
composite	carbon fibre	1.6	4300	2457
metal	aluminium	2.8	600	214
metal	stainless steel	7.86	2000	254
composite	concrete	2.3	12	4.35

^{*}The pressure needed to squash the material until it breaks.

What conclusions can you draw from the data in the table above? Q12